

Sulfur and Ammonia Calculations To Demonstrate Compliance with BAAQMD 9-1-313.2

Calculation: Refinery Fuel Gas Sulfur Removal and Recovery Efficiency on Refinery Wide Basis

Sulfur in Untreated Fuel Gas =	122	LT/D
Ideal gas volume at 60 degrees F =	380	standard cubic feet per lb-mol
Molecular weight H ₂ S =	34	
Molecular weight SO ₂ =	64	
Molecular weight COS =	60	
Molecular weight S =	32	
Long ton (LT) =	2,200	lbs

Refinery Fuel Gas	
H ₂ S daily average concentration =	10 ppmvd
Daily average flow rate =	58 MMscf/day

Sample Calculation:

**Refinery Fuel Gas
Sulfur
Emissions
(LT/D)**

$$\begin{aligned}
 &= 58 \frac{\text{MMscf gas}}{\text{day}} \times 10 \frac{\text{scf H}_2\text{S}}{\text{MMscf gas}} \times 380 \frac{\text{lb-mol}}{\text{scf H}_2\text{S}} \\
 &\times 32 \frac{\text{lb S}}{\text{lb-mol}} = 49 \frac{\text{lb S}}{\text{day}} \times 2,200 \frac{\text{LT S}}{\text{lb S}} \\
 &= 0.02 \frac{\text{LT S}}{\text{day}}
 \end{aligned}$$

Average Daily Sulfur Emissions from Refinery Fuel Gas and SCOTs

	Flow, MMscf/day	Sulfur Concentration		Total S LT/D
		H ₂ S, ppmvd	COS, ppmvd	
Refinery fuel gas	58	10		0.02
Flexigas	180	33	44	0.5
Total Refinery Fuel Gas				0.6
Total Sulfur Recovery Units (SCOTs)				0.1
54% of SCOTs S is from Refinery Fuel Gas				0.06
46% of SCOTs S is from Sour Water Strippers				0.05

Refinery Fuel Gas Sulfur Removal and Recovery Efficiency on Refinery Wide Basis

Recovered S from refinery wide fuel gas =	122	LT/D S
Emitted S from combustion of refinery wide fuel gas =	0.6	LT/D S
Emitted S from SCOT attributed to refinery wide fuel gas =	0.06	LT/D S
Total S not recovered from refinery wide fuel gas =	0.6	LT/D S
Percent total S removed and recovered from refinery wide fuel gas =	99.5%	

Calculation: Process Water Sulfur Removal and Recovery Efficiency on Refinery Wide Basis

Sulfur from Process Water to Sulfur Recovery Units = 100 LT/D S
 Sour Water = 48.3 MB/D
 Sour Water = 2,028,600 gallons per day
 HS Stripped Water Concentration = <1, below detection limit
 Molecular weight HS = 33
 Molecular weight S = 32
 Density of water = 8.34 lbs/gallon

Sulfur Remaining in Stripped Water

$$\begin{aligned}
 \text{Sulfur (LT/D)} &= 2,028,600 \frac{\text{gallons water}}{\text{day}} \times 8.34 \frac{\text{lbs water}}{\text{gallon water}} \times 0.0001\% \text{ HS, by weight} \\
 &\times \frac{32 \text{ lbs S/lb-mol}}{33 \text{ lbs HS/lb-mol}} \times \frac{1 \text{ LT}}{2,200 \text{ lbs}} = 0.007 \frac{\text{LT S}}{\text{day}}
 \end{aligned}$$

Process Water Sulfur Removal and Recovery Efficiency on Refinery Wide Basis

Recovered S from refinery wide process water =	100	LT/D S
S remaining in refinery wide stripped water =	0.007	LT/D S
Emitted S from SCOT attributed to refinery wide process water =	0.05	LT/D S
Total S not recovered from refinery wide process water =	0.06	LT/D S
Percent total S removed and recovered from refinery wide process water =	> 99.9%	

Calculation: Process Water Ammonia Removal Efficiency on Refinery Wide Basis

Nitrogen from Process Water to Sulfur Recovery Units = 79 LT/D N
 Ammonia from Process Water to Sulfur Recovery Units = 96 LT/D NH₃
 Sour Water = 48.3 MB/D
 Sour Water = 2,028,600 gallons per day
 NH₃HS Stripped Water Concentration = 50 ppmw
 Molecular weight NH₃HS = 50
 Molecular weight NH₃ = 17
 Density of water = 8.34 lbs/gallon

Ammonia in Stripped Water

Ammonia

$$\begin{aligned}
 \text{(LT/D) = } & 2,028,600 \frac{\text{gallons water}}{\text{day}} \times 8.34 \frac{\text{lbs water}}{\text{gallons water}} \times 0.005\% \text{ NH}_3\text{HS, by weight} \\
 & \times \frac{17 \text{ lbs NH}_3/\text{lb-mol}}{50 \text{ lbs NH}_3\text{HS}/\text{lb-mol}} \times \frac{\text{LT}}{2,200 \text{ lbs}} = 0.1 \frac{\text{LT NH}_3}{\text{day}}
 \end{aligned}$$

Process Water Ammonia Removal Efficiency on Refinery Wide Basis

NH ₃ from refinery wide process water destroyed in SRU =	96	LT/D S
NH ₃ remaining in refinery wide stripped water =	0.1	LT/D S
Percent total NH ₃ removed from refinery wide process water = >	99.9%	